



Chapter

# Production of Hydrogen through Gasification Technology

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## ABSTRACT

To overcome CO<sub>2</sub> emissions and environmental problems shaped due to extensive usage of fossil fuel-based feedstock, it is essential to produce potential chemicals through the cleanest energy carrier. The industrial importance of hydrogen is not needed to be emphasized. It can be obtained from biomass via thermochemical routes, which have the potential for commercial implementation as compared to other treatment routes of biomass. Biomass conversion to produce hydrogen via pyrolysis and different gasification technologies are mostly investigated. The catalytic steam reforming is attaining significant interest owing to its optimized control in catalytic and pyrolysis steps as compared to gasification. Reforming, sorption enhanced and chemical looping processes are used for removal of generated CO<sub>2</sub> (in situ) and to produce high purity H<sub>2</sub> gas. These processes are believed to shift the thermodynamic limits of WGS (water-gas shift) reaction, which enhances the conversion of feedstock and process effectiveness. The CO<sub>2</sub> capture and autothermal operating conditions can be attained through the cyclic redox mechanism of oxygen carriers (OCs) in hydrogen production by chemical looping processes. This chapter deals with biomass conversion to hydrogen through gasification technology. It also discusses the process parameters such as type of biomass feedstock types, particle size, moisture content, temperature, pressure, steam/biomass ratio, equivalence ratio and catalyst-to-sorbent ratio, which have a considerable impact on the product quality. The challenges and perspective of conversion of biomass to hydrogen are also described.